

Олимпиадная работа

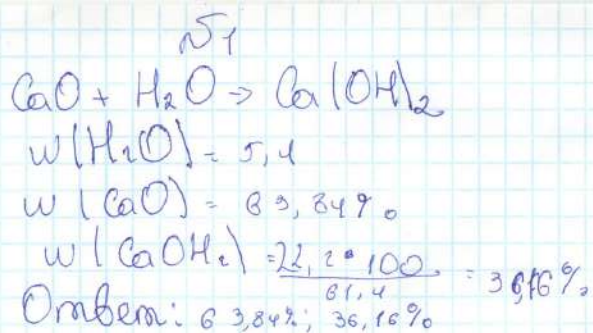
по химии

ученика 10 класса "А"

Жаровой Дарьи

285

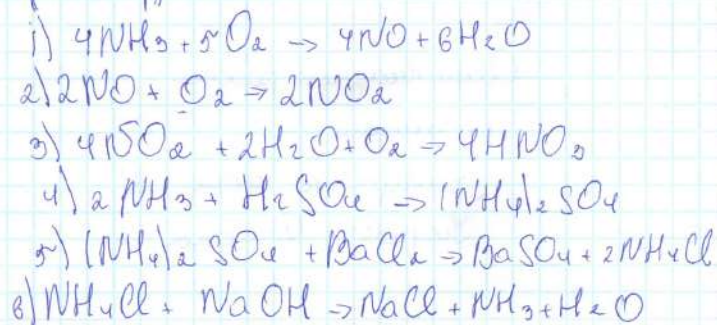
38



523

m-5.4

68



54

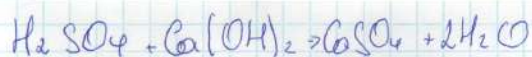
$$m(\text{H}_2\text{SO}_4) = 500 \cdot 0,88 = 440$$

$$V = \frac{440}{1,235} = 356,47$$

$$m = 140 \cdot 1,235 = 172,9$$

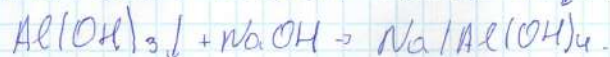
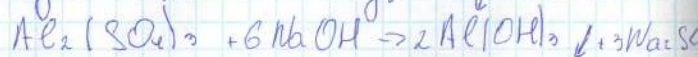
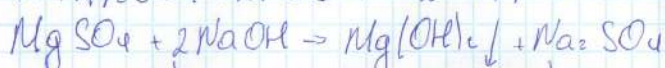
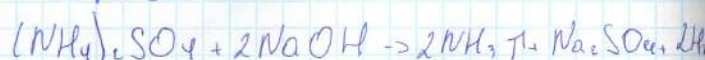
$$V(\text{H}_2\text{SO}_4) = \frac{172,9}{88} = 1,95$$

108



$$V(\text{Ca(OH)}_2) = 1,8 \cdot 74 = 133,2 \text{ l}$$

55



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